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Reducing the cost of the post-combustion CO₂ capture applied to cement plant flue gas thanks to the heat integration with a CO₂ conversion process to synthetic natural gas

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Abstract

Considering that the industrial sector contributes to around one-third of the global CO₂ emissions worldwide [1], it is mandatory to reduce the implementation costs of technologies allowing to significantly decrease the industrial CO₂ emissions, such as Carbon Capture Utilization and/or Storage (CC(U)S). Focusing on the cement industry, one of the main CO₂-emitting sectors, the post-combustion CO₂ capture process using amine(s)-based solvents is the most mature technology currently available. However, reducing its costs (both CAPEX and OPEX) is still a challenge which needs process optimization and improvements, in particular to reduce the thermal energy needed for the solvent regeneration. In parallel, Power-to-Gas (PtG) process chains using renewable H₂ and captured CO₂ as feedstocks for producing e-fuels, like synthetic natural gas (SNG), are gaining interest as complementary solution to the geological storage of CO₂, especially when CO₂ transport infrastructures are not yet installed and/or CO₂ storage sites not available near the industrial plant. Such catalytic conversion of CO₂ into SNG is generally performed in a series of adiabatic fixed bed reactors [2], and this is an exothermic process leading to the release of large amounts of usable heat. Using this excess heat as thermal source for regenerating the solvent in a CO₂ capture unit is clearly an attractive option for avoiding the use of external thermal energy and therefore reducing the operating costs of the process.

Based on these elements, the present work investigates the Aspen PlusTM (V10.0) simulation of a post-combustion CO₂ capture process applied to cement flue gases (CO₂ content of around 20 mol.%, in comparison with 5 mol.% to 15 mol.% for power plants), and its energetical integration with a catalytic methanation process using renewable hydrogen. Considering a realistic scale of 10 MW for the electrolysis power (production of 180 kg/h of H₂ and conversion of 968 kg/h of captured CO₂), as illustrated on Fig. 1, three case studies were investigated for the use of the methanation excess heat: the production of electricity [2] usable by the CO₂ capture and the methanation units (Case 1), the capture of the maximum amount of CO₂ [3] leading to some extra-CO₂ compressed in view of transport for other CCUS applications (Case 2), and the generation of steam [4] that could be used by other processes in an industrial symbiosis vision (Case 3). It is worth mentioning that the raw-SNG upgrading for grid injection is out of scope of the present work as possible use of SNG directly on the industrial plant is envisaged (a 2-stages membranes-based upgrading process example is presented in [2]). Concerning the CO₂ capture unit (see Fig. 2), based on the configurations and solvents performances comparisons performed in [2,5], it includes a Rich Vapor Compression (RVC) configuration, combined with an Inter-Cooled Absorber (ICA) and two Water-Wash (WW) sections at the top of each column. The solvent is an activated blend composed of methyldiethanolamine (MDEA) 10 wt.% and piperazine (PZ) 30 wt.%. An absorption ratio of 90 % was assumed. All the technical details regarding the CO₂ capture and the methanation units are provided in [2–4].

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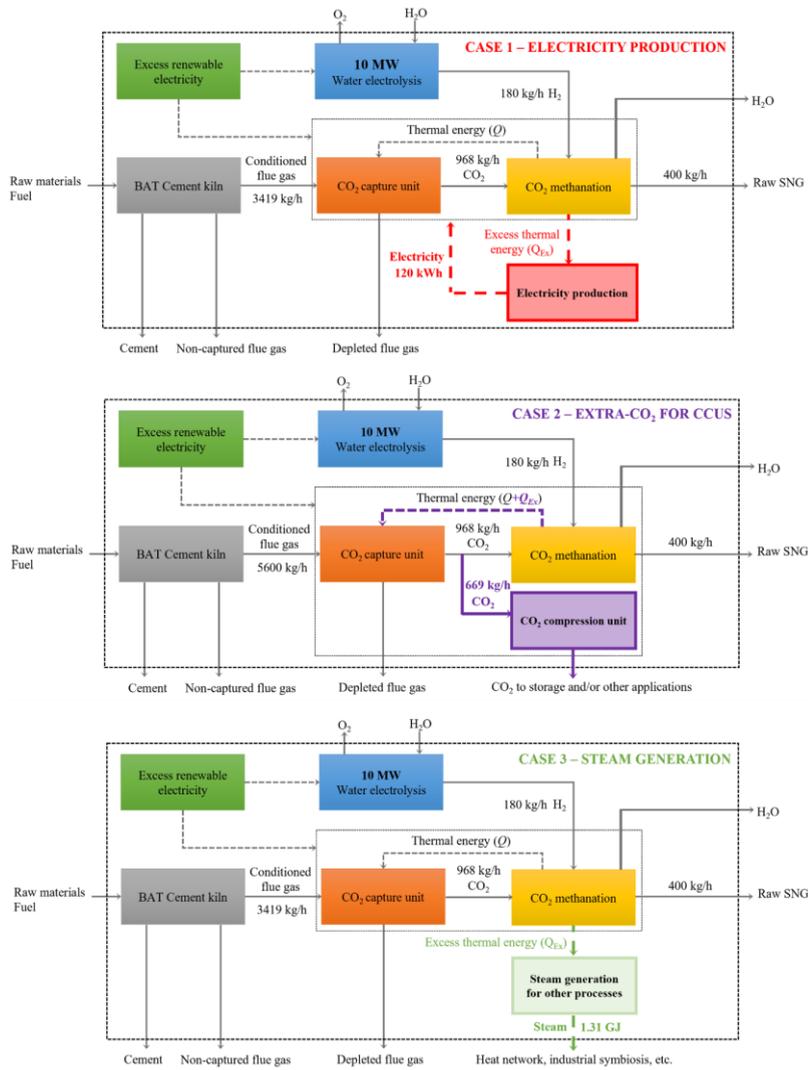


Fig. 1. Cases studies considered for the use of the excess thermal energy (Q_{Ex})

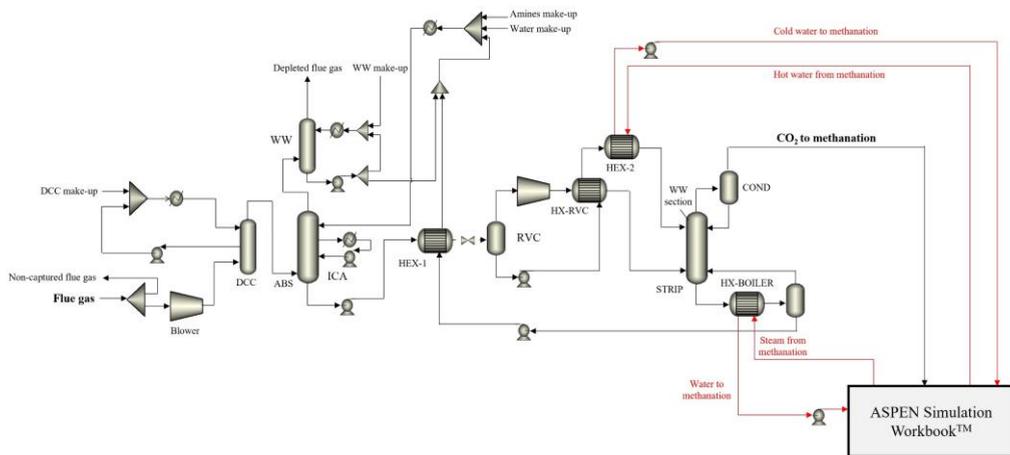


Fig. 2. Flowsheet of the CO₂ capture unit (with RVC, ICA and WW) energetically integrated with the CO₂ methanation process

Based on the simulation results, both techno-economic (TEA) and environmental (LCA) assessments were performed in order to quantify the economic and environmental interests of the proposed process chains.

Concerning the energetical results, it was shown that the implemented CO₂ capture process (RVC+ICA configuration and MDEA+PZ as solvent) leads to a Specific Reboiler Duty (SRD) of around 2 GJ/tCO₂. In addition to the configuration and solvent implemented, this low SRD value in comparison with literature ones (e.g. range of 2.3 to 3.9 GJ/tCO₂ in [6]) can be explained by the application to a cement plant flue gas (higher CO₂ content in the gas to treat in comparison with power plants, decreasing the SRD) but also by the double heat integration with the methanation unit. Indeed, as shown on Fig. 2, the hot water coming from the SNG production unit allows to give a complementary preheating to the solvent before entering the stripping column which leads also to a decrease of the SRD. Overall, in both cases, the CO₂ capture unit is self-sufficient in terms of thermal energy. Moreover, for example in Case 1, it was pointed out that in addition to the heat needed for the solvent regeneration, the electricity generated using the excess heat still available covers the entire electrical demand of the methanation unit, covering also 5% of the electricity demand of the CO₂ capture unit. On the other hand, in Case 2, the use of the entire excess heat for capturing the CO₂ allows to recover a supplementary flow of 669 kg/h of CO₂ for sequestration or other utilizations (almost 70% more than the quantity needed for the methanation unit). In Case 3, only 60% of the excess heat is used for the CO₂ capture, meaning that 40% is still available as steam for other purposes.

In terms of economics, one of the main observations is the fact that the supplementary CAPEX related to the heat integration (e.g. heat exchangers, pumps, etc.) represents only around 2% of the total CCU investment costs, while for example in Case 3 (total OPEX estimated to 117 €/tCO₂ without H₂ production) the OPEX would be 30% higher if external steam must be provided to the capture unit (without heat integration). For this Case, the OPEX and CAPEX breakdowns give a contribution of 44% related to the CO₂ capture unit and 54% to the methanation process. For the Case 2, a supplementary cost has to be considered for the CO₂ compression unit (same order of magnitude for CAPEX and OPEX as for the CO₂ capture unit itself).

Globally, the three Cases have shown an environmental interest. Different environmental impacts were investigated, including climate change (CC), fossil depletion, terrestrial acidification, freshwater eutrophication, human toxicity, water depletion, and metal/mineral depletion. Focusing on CC, in comparison with their respective reference scenario, the three approaches led to a relative decrease of the net CO₂ emissions higher than 70%. Nevertheless, the following specificities need to be considered: (i) the production of electricity needs either the establishment of, or to have access to, a high-pressure water network, as well as the implementation of a turbine, etc., which is not necessarily easy; (ii) the extra-CO₂ captured needs to fulfill the CO₂ transport network specificities in terms of purity (and such network must be available), and a potential other utilization or geological storage capacity of this CO₂ must be ensured; and finally (iii) the valorization of the excess thermal energy as steam is possible only if another industrial plant is situated not far from the cement plant itself in order to really perform an industrial symbiosis. For the three cases, it is also important to remind that the production of renewable H₂ remains an important factor for CCU, especially due to the high costs (both CAPEX and OPEX) of the electrolysis process, and also for the need of renewable electricity.

As next step of our work, the same methodologies presented in this communication could be potentially applied considering another post-combustion CO₂ capture process and/or another CO₂ conversion process.

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