

## Review Article

# Superparamagnetic iron oxide nanoparticles: promises for diagnosis and treatment of cancer

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**Abstract:** During the last decade, significant scientific research efforts have led to a significant growth in understanding of cancer at the genetic, molecular, and cellular levels providing great opportunities for diagnosis and treatment of cancer diseases. The hopes for fast cancer diagnosis and treatment were significantly increased by the entrance of nanoparticles to the medical sciences. Nanoparticles are attractive due to their unique opportunities together with negligible side effects not only in cancer therapy but also in the treatment of other ailments. Among all types of nanoparticles, surface-engineered superparamagnetic iron oxide nanoparticles (SPIONs) have been attracted a great attention for cancer therapy applications. This review covers the recent advances in the development of SPIONs together with their opportunities and challenges, as theranosis agents, in cancer treatment.

**Keywords:** Cancer, tumor, nanoscience, nanotechnology, nanoparticle, superparamagnetic iron oxide nanoparticles, SPIONs, coatings, surfaces, targeted drug delivery, controlled drug release, anticancer drugs, chemotherapy, radiotherapy, thermotherapy

## Introduction

In 2008, it has been recognized that more than 1 million of Americans and more than 10 million people worldwide were expected to be diagnosed with cancer [1]; however, in 2010 it was expected that the average percentage of people who suffer from some sorts of cancer would be 31% (according to the recent U.N. organized surveys performed in multiple countries). This considerable increase in cancer incidence may relate to the fact that determination of the cancer is strongly dependent upon the development of new diagnostic technologies. However, what is of great importance is that cancers are going to be a significant cause of human death in near future. Thus, intense scientific research was focused on the diagnosis and treatment of cancer.

Due to the dysregulated cell growth in the human body, large groups of different diseases occurred which became recognized as cancer. The uncontrollable cell-cycles not only can form

malignant tumors but also can be spread to other parts of the body through the lymphatic system or bloodstream, which is called "metastasis". Although there are numerous available anti-cancer drugs, the main problems currently associated with systemic drug administration include the general systemic distribution of therapeutic drugs, the lack of drug specificity towards a pathological site, the necessity of a large dose to achieve high local concentration, non-specific toxicity and other adverse side effects [2]. Therefore, using modern drug delivery systems are essential for high-yield cancer therapy.

The term "drug delivery" refers to the pharmaceutical agents of interest which are entrapped within, or attached to, an organic polymer matrix or inorganic particles, and in that case, drug safety and efficacy can be greatly improved and new targeted therapies are possible [2]. The strategy for drug delivery can be approached by many methods: (1) drug modification by chemical means [3, 4]; (2) drug encapsulated in vesi-

cles [5-7] or inorganic hollow nanoshells [8-10]; (3) controlled release system such as polymer-based [11-13] and hydrogel-based vectors [14-16], and (4) small-scale integrated systems such as microchip systems [17-20]. The delivery routes are also an important issue for systemic delivery of drugs due to the systemic circulation. Intravenous [21, 22], intraperitoneal [23], pulmonary [24], transdermal [25-27], nose [28, 29], vaginal [30] and eye [31, 32] routes of administration are the approaches designed to improve the selectivity of chemotherapy.

Although there are significant improvements in targeted drug delivery systems, there is a great deal of commonly pursued, desirable qualities for such efficient drug delivery devices, which are not necessarily guidelines. The crucial areas of potential that powerful drug delivery systems should have are: (1) long circulation (i.e. long half-life), (2) high levels of bioavailability and specific targeting, (3) intracellular/organelle delivery, (4) stimuli responsiveness (i.e. efficacy), (5) reporting/imaging, and (6) biodegradation.

In recent years, the fabrication of nanoparticles and exploration of their unusual properties have attracted the attention of physicists, chemists, biologists and engineers. Interest in nanoparticles arises from the fact that the mechanical, chemical, electrical, optical, magnetic, electro-optical and magneto-optical properties of these particles are different from their bulk properties and depend on the particle size. Entrance of nanoparticles in the medical field caused significant hopes for early diagnosis and treatment of catastrophic diseases such as cancer. Nanoparticles are typically referred to as microscopic particles that are between 1 nm and 100 nm in size. Using nanoparticles, delivery systems have the potential to target drugs to specific sites of the body or precisely control drug release rates for prolonged times according to body reaction of chemicals microenvironment [2]. More specifically, nanoparticles as drug carriers in targeted delivery systems in cancer therapies can provide desired and precise penetration of therapeutic and diagnostic substances within for example tumor sites/tissues/cells while their corresponding side effects are much lower in comparison to conventional cancer therapies [2].

Nanoparticles, as smart drug delivery systems,

have strong capability to overcome numerous challenges that still exist including the targeting of drugs to specific cells, the creation of novel vaccine delivery approaches, and the development of cell-based delivery systems. Nanoscale drug delivery devices that are stimuli responsive respond dynamically to changes in the environment. These stimuli could be in the form of changes in temperature [33], light [16], pH [15, 34, 35], ultrasound [35-37], or magnetic fields [38, 39]. Disease processes which upset homeostasis can lead to environmental changes that can be exploited by stimuli-responsive therapeutics. Thus, in recent years, significant efforts have been devoted to develop nanotechnology for drug delivery since it offers a suitable means of delivering small molecular weight drugs [40], as well as macromolecules such as proteins, peptides [41] or genes [42, 43] by either localized or targeted delivery to the tissue of interest. It is also worthwhile to mention that nanoparticles may have promise for biomolecular imaging [44]; thus, another potentially important facet of nanoparticles and its application in cancer research could be its capability to probe underlying mechanisms in cancer.

In order to increase the efficiency of drug delivery, magnetic nanoparticles are defined as a promising candidate not only due to capacity of antibodies to attach to their surfaces, but also due to achieve targeting ability by using external magnetic guidance [45, 46]. Among different types of magnetic particles, superparamagnetic iron oxide nanoparticles (SPIONs), with a mean diameter as low as 10 nm and superior magnetic properties, have proven to be among the most capable candidates [47]. In the field of drug delivery, SPIONs particles are considered as small, thermally agitated magnets in carrier liquids, which are called "ferrofluids". A distinguishing feature of SPIONs for drug delivery is their applicability for both alternatives (i.e. magnetic properties and antibody attachment) and consequently developing a targeting capability [2]. Superparamagnetism essentially acts as an activation mechanism because once the external magnetic field is removed, the magnetization disappears, and thus the agglomeration, and hence the possible embolization of the capillary vessels can be avoided [48].

SPIONs also have capability to be used as multifunctional agents [49-51]. The surfaces of SPIONs are engineered to have several simultane-

**Table 1.** Principal preparation methods of SPIONs

Synthetic method	Advantages	Disadvantages	References
Coprecipitation	Rapid synthesis with high yield	Problem of oxidation and aggregation	[48, 62]
Hydrothermal reactions	Narrow size distribution and good control, scalable	Long reaction times	[63, 64]
High temperature decomposition	Good control of size and shape, High yield	Further steps needed to obtain water stable suspension	[61, 65, 66]
microemulsion	Control of particle size	Poor yield and large amounts of solvent required, excess of surfactant to eliminate	[51, 67, 68]

ous biomedical-related functions such as drug carrier properties, magnetic resonance imaging (MRI) contrast agents, and local heat induction (hyperthermia) capacity. More recently, a field called “molecular imaging” has appeared, this technique allows the in vivo visualization of molecular events occurring at the cellular level and needs the development of high affinity ligands and their grafting to SPIONs [44]. SPIONs are also good substrates for bioconjugation, they are used as reporters for many physiologic processes and have a lot of clinical applications such as liver and spleen imaging, inflammation, apoptosis, and cardiovascular disease [45, 52, 53].

SPIONs can be accumulated in a biological site by passive or active targeting mechanism. In passive targeting, contrast agents are concentrated in the phagocytic cells (e.g. Kupffer’s cells and macrophages) and corresponding organs (e.g. liver, spleen, and lymph nodes) that are responsible for clearance from the body. For example, SPIONs coated with dextran, carboxy-dextran have been used for passive targeting [54-56]. In active targeting, contrast agents need conjugation to a specific ligand to target a site of interest. Typical ligands include antibodies, peptides, sugars, aptamers, etc. which are molecules that can be linked to SPIONs covalently or non-covalently [52, 57, 58].

For biological applications, nanoparticles must be highly stable in aqueous ionic solutions at physiological pH. Vectors grafted on their surface must be able to recognize the target cells or tissues. Particles must be non-toxic and remain in the circulation for a sufficient amount of time to allow targets to be reached. Intensive research are currently undertaken to develop specific contrast agents in targeted cancer imaging. For example,  $\alpha_v\beta_3$  integrin targeted SPIONs have been used for the specific MRI detec-

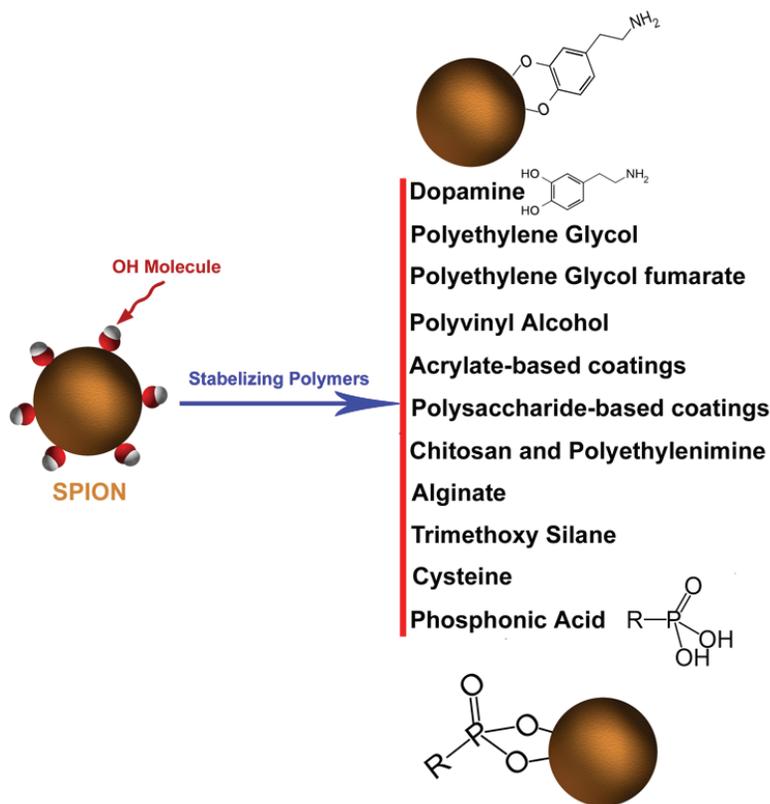
tion of small regions of angiogenesis associated with solid tumors [59, 60].

In summary, SPIONs can be used as theranosis agent in the context of cancer, in other words it can serve as both a diagnostic and therapeutic agent. In this review, the main protocols for the grafting of vectors to SPIONs are reported and some applications are described. In addition, we discuss various in vitro and in vivo studies using surface engineered SPIONs for diagnosis and treatment of cancer diseases.

### Synthesis and stabilization of SPIONs

Several methods for chemical synthesis of SPIONs have been described. The most commonly used are summarized in **Table 1**. Amongst these methods, co-precipitation of  $Fe^{2+}$  and  $Fe^{3+}$  ions in a basic aqueous media (e.g. NaOH or  $NH_4OH$  solutions) is the simplest way, but usually nanoparticles are polydispersed and poorly crystallized [48]. To avoid these disadvantages, thermal decomposition methods have been employed to produce SPIONs with monodispersity and uniform crystalline [61]. Subsequently, the hydrophobic iron oxide nanoparticles can be coated with phospholipids, silica, or amphiphilic polymers as shells to display good solubility and biocompatibility in vivo (**Figure 1**).

SPIONs can be coated in situ during the nucleation and growth of the magnetic core (this simultaneous process is often referred to as the “one-pot” method) or after the synthesis following the final application. Monomeric organic stabilizers, polymers and inorganic coatings can also be used to stabilize nanoparticles in aqueous solutions [46, 69]. Organic surfactants are frequently used for the stabilization and coating of magnetic nanoparticles. Fatty acids can stabilize the aqueous fluids by the formation of a surface bilayer with a chemisorbed fatty acid



**Figure 1.** Scheme showing representative groups that can be used to stabilize the SPIONs.

primary layer and an interpenetrating second layer, the latter is physisorbed onto the primary layer with the hydrophilic head-groups pointing outwards [70].

Coating agents which are physically adsorbed (by electrostatic interactions or hydrogen binding) show limited stability in comparison to coating agents which are chemically adsorbed. The stability of the coating grafting also depends on the quantity of the chemical interaction that each molecule or macromolecule can establish with the SPION surface. The most common coatings for biocompatible iron oxide suspensions are polymers [71] such as derivatives of dextran [72, 73] (dextran, carboxymethylated dextran, carboxydextran), arabinogalactan, glycosaminoglycan, starch, polyethyleneglycol, siloxane, sulphonated styrene-divinylbenzene, poly(lactic acid), poly( $\epsilon$ -caprolactone) or polyalkylcyanoacrylate [74-76].

Due to their good solubility in water and biocompatibility, polysaccharides are among the most

commonly used coating for the stabilisation of SPIONs. Dextran-SPION can be prepared using co-precipitation method with in situ coating by polysaccharide [77]. Molday and MacKenzie used ferrous and ferric chloride under basic condition in the presence of dextran. Dextran-coated SPIONs with surface functionalities were also developed for specific applications: carboxydextran for cell labeling [78] or aminodextran for grafting with DNA [79]. The chemical modifications of dextran have showed that reduction of terminal sugars can have a significant effect on particle size and coating stability [80]. Particles prepared with carboxydextran yielded a more stable coating [71, 81]. Considering the dextran-coated particles, no evidence of strong chemical adsorption was observed by FTIR (Fourier transform infrared spectroscopy) and SSIMS (statistic secondary ion mass spectra) analysis [82]. The nature of the interactions between the dextran and the SPION surface

and its evolution with temperature has been investigated by thermogravimetric and differential thermal analyses and by coupling these data with FTIR analysis [83]. Noteworthy, these dextran-coated iron oxide particles do not show any toxicity [84-87].

Other surface-modifying agents have been explored to increase stability of magnetic nanoparticles. In order to obtain a strong conjugation of dextran to the maghemite surface, Mornet et al. [88] have described a synthetic route consisting of surface modification of SPIONs by silanation of the iron core with aminopropylsilane groups and covalent conjugation with partially oxidized dextran and subsequent reduction of the Schiff base [89].

The bonding nature of organosilanes to iron surfaces can be analyzed by FTIR and secondary ion mass spectroscopy (ToF-SIMS). The Si-O-Fe bond is commonly described as covalent [90]. Such systems are ordered molecular assemblies formed by the adsorption of an active

molecule (e.g. siloxane, carboxylates, thiolates, and phosphate) on a solid surface (as iron oxide) with different terminal groups (usually -OH, -COOH, and -NH) [91, 92]. This coating can present terminal groups allowing further functionalization by chemical reactions. These functionalities are also frequently incorporated in various polymers where attachment of species on the surface is desired. The stability depends basically on the affinity of the active molecule for the substrate (solid surface), pH and ionic strength of the environment.

Dextran-coated superparamagnetic iron oxide particles can also form stable complexes with transfection agents. Moreover, such complexes can be internalized by endosomes/lysosomes, and have been utilized for tell labeling and in vivo MRI tell tracking [93]. Alginate, another polysaccharide, has also been used for the coating and stabilization of magnetic nanoparticles [51]. Starch-coated SPIONs, obtained via co-precipitation in the presence of starch were also investigated for targeting of brain tumors in rats [94]. Chitosan, a biocompatible and biodegradable polymer, is of particular interest for coating magnetic nanoparticles [95, 96]. It has been reported that oleic acid-coated SPIONs can be easily dispersed in chitosan, producing stable ferrofluids with a typical hydrodynamic diameter of approximately 65 nm [97].

Poly(lactic acid), another biodegradable polymer, has been used to prepare stable biocompatible ferrofluids with different ferromagnetic particle sizes, ranging from 10 to 180 nm [98]. Poly(lactic acid)-coated nanoparticles can also be loaded with anticancer drugs (e.g., tamoxifen), which allows their use in simultaneous tumor imaging, drug delivery and the real-time monitoring of therapeutic effects [99]. Similar biocompatible nanoparticles have also been prepared via an in situ controlled co-precipitation of magnetite from aqueous solutions containing suitable  $\text{Fe}^{2+}$  and  $\text{Fe}^{3+}$  salts, in a polymeric starch matrix. This process resulted in starch-coated SPIONs that demonstrated good potential for the imaging of nerve tells and the brain [94].

One very successful strategy for the preparation of stable and biocompatible nanoparticles is to graft poly(ethylene glycol) (PEG) onto the surface (a process known as PEGylation). PEG is not only biocompatible but also has favorable chemi-

cal properties and solubility. In this situation, the stabilization is due primarily to steric interactions, while PEGylation can be used to further enhance the pharmacokinetic properties and improve the blood circulation times [100, 101]. For example, an increased image contrast in MRI was achieved by using polymeric micelles formed from SPIONs encapsulated in biocompatible, biodegradable poly( $\epsilon$ -caprolactone)- $\beta$ -PEG copolymers. These materials have demonstrated significantly improved  $r_2$  relaxivities and a good sensitive MRI detection [102]. Copolymers of PEG were also used to stabilize the SPIONs [51, 62].

Aside from the extended half-life that it can provide, one of the great advantages of PEG coating is that it can also be easily conjugated to antibodies or to peptides to achieve a specific targeted delivery. For example, in a recent report, biocompatible water-soluble magnetite nanocrystals were fabricated via the thermal decomposition of ferric triacetylacetonate in 2-pyrrolidone in the presence of monocarboxyl-terminated PEG (MPEG-COOH) [103]. The carboxylic acid groups on the surface of the particles were conjugated with a cancer-targeting anti-carcinoembryonic antigen (CEA) monoclonal antibody, via a carbodiimide coupling reaction. The resultant materials were assessed for their ability to label cancer tissues in vivo, for subsequent MRI detection [2]. PEG-coated iron oxide nanoparticles may also be conjugated to specific targeting peptides and receptors such as chlorotoxin [104], transactivator protein (Tat) of HIV-1 [105], and integrins [59].

The high temperature decomposition process, known to give nanoparticles a better control of particle size and monodispersity, was described as a one-pot method by decomposition of  $\text{Fe}(\text{acac})_3$  in 2-pyrrolidone in the presence of MPEG-COOH [103]. Another way to introduce PEG onto SPIONs is to use PEG-silane [106]. This involves reaction of the triethoxysilane group with the hydroxyl group of the SPION surface. Sun et al showed that SPIONs can be treated successively with a bifunctional silane PEG trifluoroethylester linker, ethylenediamine and folic acid (FA). The specificity of these vectorized SPIONs was demonstrated by an increased nanoparticle uptake and a significant contrast enhancement of tumor cells. An alternative protocol to prepare SPION-PEG-FA in-

volved the reaction of FA with tertbutyl-oxycarbonyl (t-BOC) and N-hydroxysuccinimide (NHS), followed by the reaction of (t-boc)folate-NHS with amino-PEG-carboxyl [107]. The resultant (t-boc)folate-PEG-carboxyl was then reacted with SPION with have NH<sub>2</sub> groups obtained by reaction with 3-aminopropyl-trimethoxysilane. Finally, the t-boc protective group was removed. Recently, another anchor, dopamine (DPA) has been proposed due to its high affinity for the iron oxide nanoparticle surface and the possibility of functionalization with other molecules through amide bonds [108].

Other polymers and copolymers, which have been used to coat magnetic nanoparticles, include PVP), polyethylenimine (PEI), polyvinyl alcohol (PVA), polysodium-4-styrene sulfonate, poly(trimethylammonium ethylacrylate methyl sulfate)-poly-(acrylamide), polyvinylbenzyl-O-beta-D-galactopyranosyl D-gluconamide (PVLA), polycaprolactone, and gummic acid [51, 62]. In addition, several stable and biocompatible magnetic fluids have been prepared by coating magnetic nanoparticles with proteins, such as human serum albumin (HSA), avidin, and Annexin AS (anxA5)-VSOP [109, 110].

Dicarboxylic or tricarboxylic acids (citric, tartaric or dimercaptosuccinic acids) [111, 112] are also used for the surface functionalization and stabilization of SPIONs. Some of the functional groups can bind to the surface of the iron oxide, while the remaining carboxylate groups provide negative charges (depending on the pH) and improve the hydrophilicity of the SPION surface. Several studies are reported and demonstrated that phosphonates and phosphates bind efficiently to iron oxide particle surfaces and can serve as potential alternatives to fatty acids [92, 113, 114]. Functionalized phosphonate and phosphate seems to have an acceptable biocompatibility [115] and it is possible to suggest their utilization as coating agents of magnetic nanoparticles in medical applications [116]. Phosphonates are molecules that contain one or more R-PO(OH)<sub>2</sub> Lewis acid groups. The P-C bond is very stable toward oxidation or hydrolysis. These compounds possess a very high ability to form strong complexes with transition metals in aqueous solution and show a large affinity for the metal oxide surfaces [117].

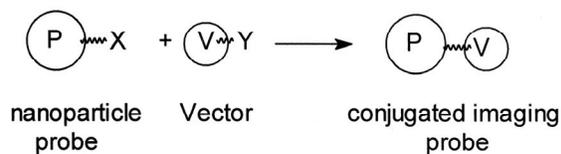
The mechanism of adsorption of dimercaptosuccinic acid (DMSA) has been studied by con-

ductimetric measurements and adsorption isotherms curves. DMSA is oxidized during the coating process in tetrameric polysulfide chains [DMSAox]<sub>4</sub> which are absorbed by the carboxylate moiety on the particles after alkylation and neutralisation. The obtained particles are stable particles at pH = 7 [118, 119].

Among the inorganic coatings, silica, carbon, precious metals (e.g., Ag and Au), or metal oxides are the most frequently used [69]. The silica coating significantly improves the stability of magnetic nanoparticles, protecting them from oxidation, and reduces any potential toxic effects of the nanoparticles [62]. Silica coating can be achieved by using several different approaches; the most popular is the sol-gel process with tetraethyl orthosilicate (TEOS) (known as the Stober method) [120, 121]. Silica shell formation is achieved by the hydrolysis of TEOS in the presence of ammonia and SPIONs, the thickness of the silica coating can be controlled by varying the concentration of ammonia and the ratio of TEOS to water. Aminosilane coatings were activated using glutaraldehyde, which served as a linker for the binding of a monoclonal antibody directed against cancer. This process resulted in new immunomagnetic nanoparticles for the targeted MRI of cancer [122].

### Strategies of vectorization of magnetic nanomaterials for targeted imaging

Targeted cellular labeling and molecular imaging require further functionalization in order to provide molecular recognition for specific biological sites. Vectorization is also critical for the further stabilization of nanoparticles, to improve their biocompatibility, and to reduce their potential toxicity. The main vectorization strategies include: (i) the noncovalent grafting of biomolecules (e.g., antibodies or proteins) via ionic bonding or adsorption; and (ii) the covalent conjugation of biomolecules via strong chemical bonding [123]. Typical examples of the noncovalent approach include the preparation of streptavidin-coated iron oxide nanoparticles [124-126]. Although the noncovalent methods are relatively easy to undertake, the results are very often not reproducible and the response of the materials may be very difficult to control. In addition, noncovalently functionalized nanocomposites are sometimes unstable in variable biological media, and may



**Figure 2.** Covalent grafting of vectors (V) onto the nanoparticles (P) such as for a molecular imaging probe.

lose their biological coating and undergo precipitation. Therefore the development of a covalent approach has attracted much more attention during recent years. This involves the formation of linkages between vectors and nanoparticles (see **Figure 2**).

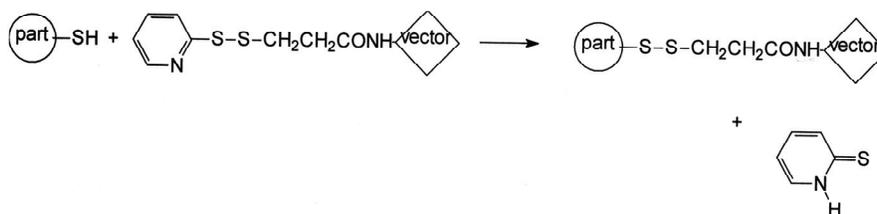
The linkage must be stable (it should be resistant to hydrolysis, oxidizing and reducing conditions) and performed in mild conditions to avoid vector degradation. Typical vectors are peptides, proteins, antibodies, mimetic molecules, carbohydrates, lipids, etc. The usual reacting group is an amine or a carboxylic function in the presence of carbodiimine as activator of the carboxylic group but reactions with hydroxyl, thiol or phenol residues are also possible.

An oxidative conjugation strategy has been used in previous studies. This method is based on the periodate oxidation of a carbohydrate polymer like dextran or carboxydextran to aldehydes, which may then be linked to biomolecules through the formation of a Schiff base. This strategy has been used for the covalent conjugation of dextran coated magnetic nanoparticles with peptides [127], proteins [128, 129], monoclonal antibodies [130-132] or agglutinin [133]. A substantial loss of the biological activity of the protein has been observed. To minimize this effect, Hogemann et al. [134] have linked proteins and iron oxide particles via a linker. Their results suggest that the oxidative

conjugation chemistry significantly interferes with the binding of the conjugates of the receptor. Current efforts are devoted in the direction of non oxidative strategies. The target molecules can be covalently linked through a 3 step-reaction sequence described by Josephson et al. [135]. This approach is based on amine-terminated CLIO nanoparticles, which can be obtained from dextran-coated nanoparticles by cross-linking using epichlorohydrin and then ammonia. A peptide was attached to the amino group of a cross linked dextran iron oxide (CLIO-NH<sub>2</sub>) using SPDP through a disulfide exchange reaction. A range of target biomolecules can be conjugated by using standard organic chemistry methods: formation of disulfide, carbonthiol, and amide bonds [105, 136-139].

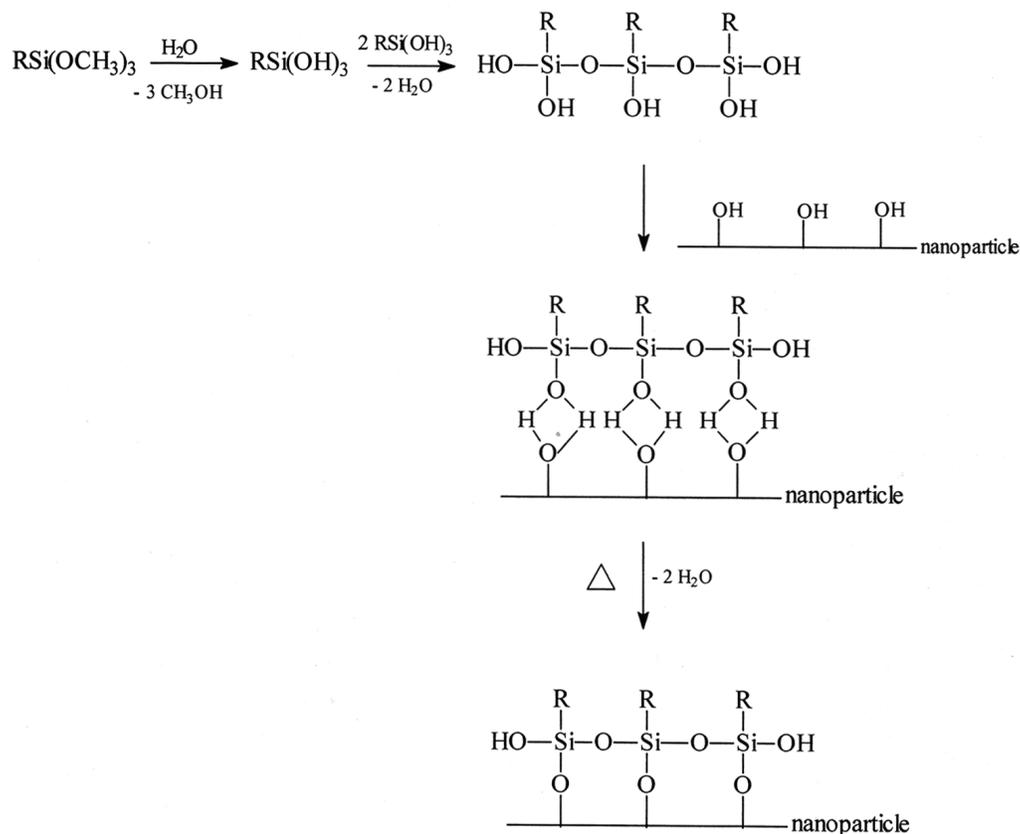
In the so-called “DMSA techniques” 2,3-dimercaptosuccinic acid (DMSA) -coated magnetic nanoparticles can be covalently linked to a variety of biomolecules via S-S bonds using N-succinimidyl 3-(2-pyridyldithio) propionate (SPDP) as a coupling agent (see **Figure 3**) [118]. This approach has been used to couple antibodies, lectins and annexin V to DMSA-coated magnetic nanoparticles [119, 140, 141].

Magnetite nanoparticles coated with silica were prepared, after surface modification with an amino-silane coupling agent, SG-Si900, amine was covalently linked using glutaraldehyde as a cross-linker [142]. Alternatively, vectors with carboxylic functions can be directly grafted on the silica coated particles using EDC to activate the carboxyl groups [143]. The silane coupling materials (like 3-aminopropyltrimethoxysilane or p-aminophenyl trimethoxysilane) [144] are able to adsorptively or covalently bind to the metal oxide and are able to form covalent bonds with bioaffinity adsorbents through organo-functionalities. The mechanism of the silane coupling agents reaction according to Arkles



**Figure 3.** Formation of grafted particles via S-S bridge: the pyridyl sulfide moiety of SPDP grafted on the vector is substituted by the SH group on the nanoparticles. Reprinted with permission from (reference 52). Copyright 2010 American Chemical Society.

## Superparamagnetic iron oxide nanoparticles and cancer



**Figure 4.** Adsorption mechanism of organosilane and chemical reactions of silane coupling agents on magnetic particles (R= CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>NH<sub>2</sub>, CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>CN, ...; R'=CH<sub>3</sub>, C<sub>2</sub>H<sub>5</sub>). Reprinted with permission from (reference 52). Copyright 2010 American Chemical Society.

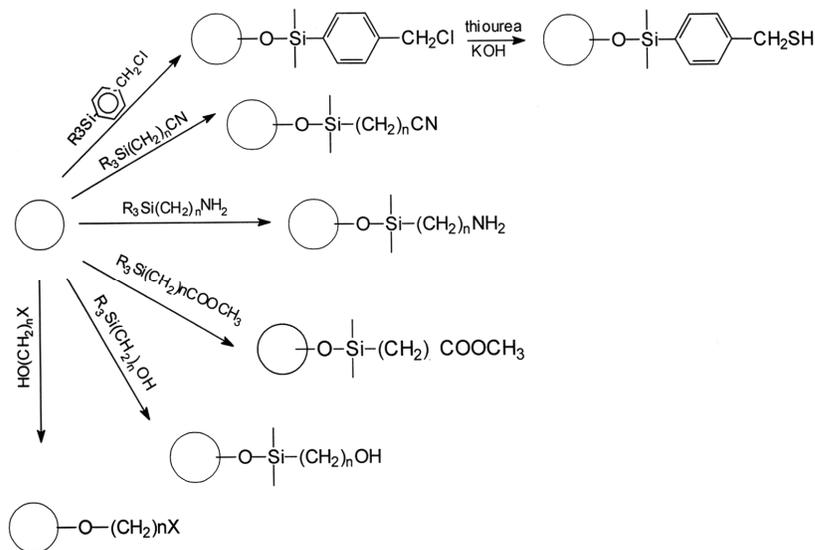
was depicted in **Figure 4** [145].

The silane is deposited on the metal oxide core from acidic solution (see **Figure 5**). The silinization reaction occurs in two steps: (i) the trimethoxysilane is placed in acidic water, phosphorous acid and glacial acetic acid and it condenses to form silane polymers; (ii) these polymers associate with the metal oxide by forming covalent bond with surface OH groups through dehydration or by adsorption of silane polymers to the metal oxide. Diazotation of aminophenyl-terminated silane or the use of glutaraldehyde on 3-aminopropyl-terminated silane can be used to couple antibodies or immunoglobulins. This second procedure also consists of two basic steps: (i) activation of the particle by reaction with glutaraldehyde followed by (ii) removal of unreacted glutaraldehyde and reaction of the proteins with the activated particles followed by removal of the unreacted proteins. If the magnetic particles are coated by carboxy-terminated silanes, proteins can be coupled to them by

treating the particles with 3-(3-dimethylaminopropyl) carbodiimide. The surface chemistry involving reactions with alkyltrialkoxysilane or trichloroalkylsilane compounds is a good way for grafting biomolecules [146, 147].

Finally, the recently developed “click” chemistry, based on the azide-alkyne reaction, has been applied to the functionalization of iron oxide nanoparticles [114, 148, 149], and allows the relatively simple synthesis of azido- or alkyne-functionalized nanoparticles, which then can be linked to appropriate target molecules. For example, Turro et al. [114] described the stabilization of Fe<sub>2</sub>O<sub>3</sub> nanoparticles using alkyne-terminated organophosphate or carboxylic acid groups to exchange with oleic acid on the Fe<sub>2</sub>O<sub>3</sub> surface. The IONPs were subsequently covalently attached to poly(tert-butyl acrylate) via click reactions using CuSO<sub>4</sub>.

A pan-bombesin analog was conjugated through a linker to dye-functionalized superparamag-



**Figure 5.** Possibilities of surface modification with organosilane following the chemical reactions described in Figure 4. Reprinted with permission from (reference 52). Copyright 2010 American Chemical Society.

netic iron oxide nanoparticles for the targeting of prostate cancer cells. The peptide was conjugated via click chemistry. The peptide-functionalized nanoparticles were then demonstrated to be selectively taken up by PC-3 prostate cancer cells relative to unfunctionalized nanoparticles and this uptake was inhibited by the presence of free peptide, confirming the specificity of the interaction [150].

### SPIONs against cancer diseases

#### Drug delivery

Widder et al. [151] employed the first magnetic drug (i.e. doxorubicin (Dox)) delivery systems, where Dox was encapsulated in albumin magnetic nanoparticles. After this report, intense research has focused on targeted delivery and imaging using magnetic nanoparticles [152-169]. Some of the developed magnetic nanoparticles have rapid hepatic uptake after intravenous administration; these uptakes would be of crucial importance for hepatic tumors diagnosis and treatment [170, 171]. However, the crucial matter for high-yield drug delivery using magnetic nanoparticles is their targeting capabilities. In order to increase the targeting capability of magnetic particles, it is essential to attach targeting moieties (e.g. antibodies and hormones) to the surface of magnetic nanoparticles. In this case, there are varieties of methods for conjugation of antibodies to the surface of magnetic nanoparticles with no detectable effects on the colloidal stability of the

particles [172, 173]. Attachment of epidermal growth factor (EGF) to the magnetic nanoparticles can be useful in treatment of colorectal and breast cancers [174]. HER2 antibody was also conjugated with glycerol mono-oleate coated SPIONs and the resulting materials showed enhanced uptake in human breast carcinoma cell line (i.e. MCF-7) [175]. Hormones (e.g. LHRH) conjugated SPIONs also showed good capability for targeting of cancer cells in both the primary breast tumors and the lung metastases cells [176].

In relation to targeting properties, there are also two main additional problems for high yield targeted delivery. First, as the drug coats the surface of nanoparticles, a significant portion of it is quickly released upon injection (i.e. burst effect). Therefore, only small amounts of the drug reach the specific site after, for instance, magnetic drug targeting. Second, once the surface-derivatized nanoparticles are inside the cells, the coating is likely digested, leaving the bare particles exposed to other cellular components and organelles, thereby potentially influencing the overall integrity of the cells. To overcome these two shortcomings, Mahmoudi et al. [177] used cross-linked poly (ethylene glycol)-co-fumarate (PEGF) coating on the surface of SPIONs. To investigate if the coating could reduce the burst effect, nanoparticles were prepared by incorporating the anticancer drug Tamoxifen (TMX). The cross-linked PEGF coating reduced the burst effect rate by 21% in comparison with the noncross-linked tamoxifen nanoparticles

[177]. Thus, the authors claimed that the SPIONs with coating based on crosslinked unsaturated aliphatic polyesters are potentially useful to develop novel carriers for drug and gene delivery applications [177].

Another crucial matter is the multifunctional capabilities of the engineered SPIONs, which are a key focus in bionanotechnology and will have profound impact on molecular diagnostics, imaging and therapeutic of cancer diseases. Gao et al. [178] prepared FePt coated SPIONs and claimed that their core-shell particles have good capability to be functionalized by various targeting molecules (e.g. antibodies). These multifunctional nanoparticles can enhance the capability of magnetic particles for simultaneous detection and monitoring of the transformation of tumors by noninvasive MRI during chemotherapeutic treatment by nanoparticles. In order to combine multiple components on a nanometer scale for creation of new imaging modalities, which are unavailable from individual components, Jin et al. [179] prepared multifunctional nanoprobcs. These not only offered contrast for electron microscopy, MRI and scattering-based imaging but also, more importantly, enabled a new imaging mode, magneto-motive photoacoustic imaging. This form of imaging with remarkable contrast enhancement compared with photoacoustic images using conventional nanoparticle contrast agents; these particles were composed of thin gold coated SPIONs by creating a gap between the core and the shell. Very recently, Mahmoudi et al. [50] synthesized a multi-component system made of gold-coupled core-shell SPIONs, as a new nanoprobe with signal enhancement in surface Raman spectroscopy, due to its jagged-shaped gold shell coating. These new classes of multifunctional magnetic nanoparticles [50, 179] may have great impact in the future in relation to cancer diagnosis and therapy.

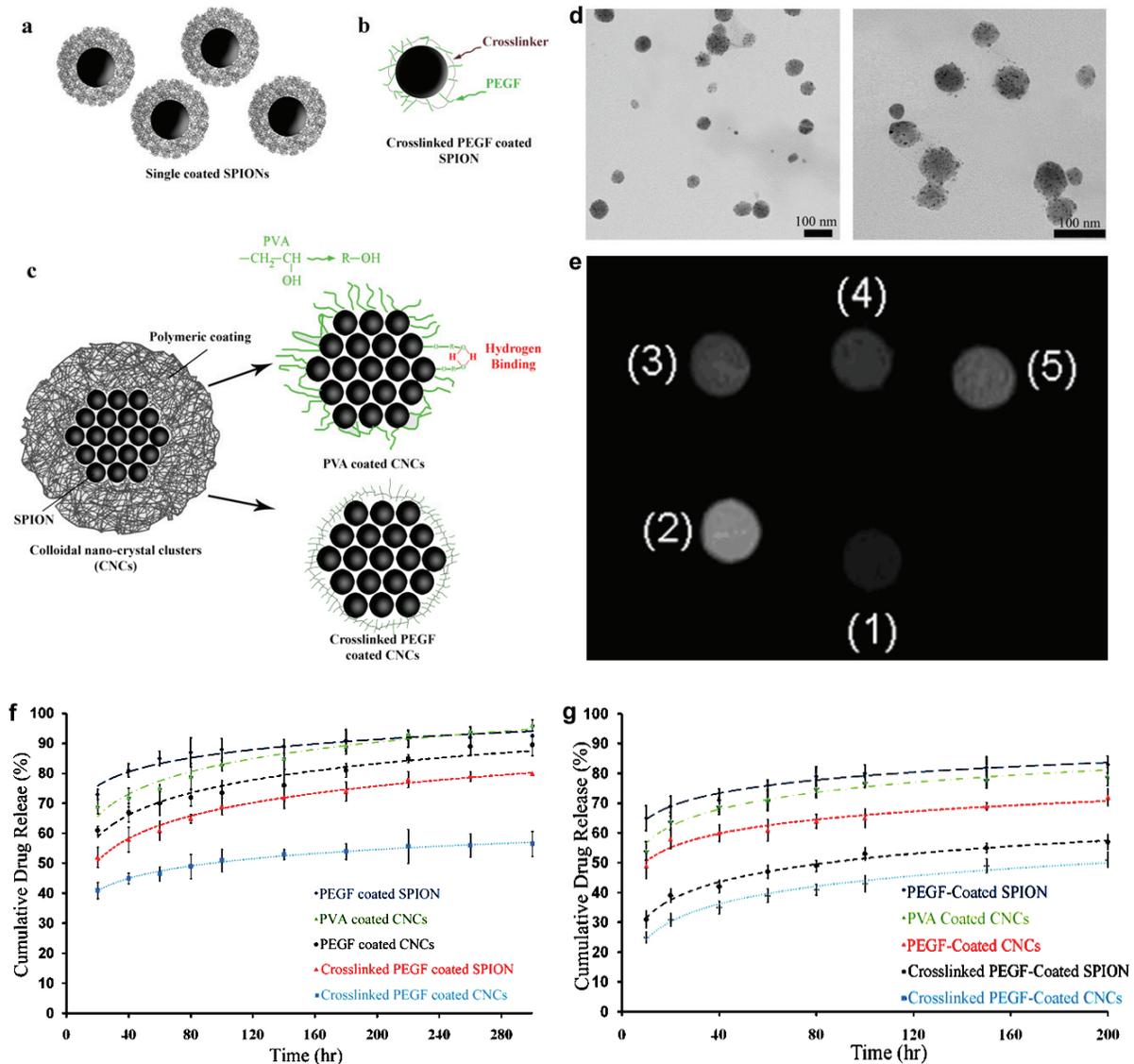
Amiri et al. [180] reported cell endocytosis, drug release, NMR relaxometry and in vitro MRI studies on a novel class of superparamagnetic colloidal nano-crystal clusters (CNCs) with various biocompatible coatings. They claimed that the transverse relaxivity  $r_2$ , the parameter representing the MRI efficiency in negative contrast agents, for the polyvinyl alcohol (PVA)-coated, polyethylene glycol-co-fumarate (PEGF)-coated, and crosslinked PEGF-coated CNCs was efficient enough to contrast suitably the MRI (see

**Figure 6**). In addition, their prepared samples have been shown to facilitate controlled drug release (particularly the crosslinked PEGF-coated compound), thereby finally allowing them to propose this class of compounds for future applications in cancer, as theranostics agents.

Besides in vitro evaluations of drug loaded surface engineered magnetic nanoparticles, there are numerous in vivo targeting assessments. For instance SPIONs have been injected very close to tumor sites in animal models and human clinical studies for targeting delivery of anticancer drugs [181-188]. SPIONs have the capability to enhance MRI in combination with diffusion weighted MRI, which can be a novel, accurate, and fast method for detecting pelvic lymph node metastases even in normal-sized nodes of patients with bladder or prostate cancer [189]. The surface engineered SPIONs can be also employed in diagnosis and treatment of colorectal cancer [190].

In order to increase the yield of magnetic targeting, Widder et al. [191, 192] employed intra-arterial injection proximal to the tumor site (Dox filled magnetic particles). Their results demonstrated a 200-fold increased targeting yield in comparison with intravenous injection [193]. Since this study, success in cytotoxic drug delivery and tumor remission has been reported by several groups using animals models including swine [194, 195], rabbits [196] and rats [197-199]. In order to increase the targeting yield of magnetic nanoparticles, permanent magnets can be implanted in a targeted site; for example, permanent magnets were implanted at solid osteosarcoma sites in hamsters and anticancer drugs subsequently delivered to the targeted site using magnetic liposomes [200]. Using magnetic implantation methods, the yield of targeting were enhanced four-fold in comparison with normal intravenous (non-magnetic) delivery. Results also showed significant increase in anti-tumor activity and the elimination of weight-loss as a side effect [201]. This technique has also been employed to target cytotoxic drugs to brain tumors, where passing the drugs through the blood-brain barrier is difficult [199]. Preliminary successful animal trials have lead to the development of magnetic nanoparticles for use in human trials. For example, magnetic nanoparticles (metallic Fe coated with activated carbon) which carried the Dox as drug

## Superparamagnetic iron oxide nanoparticles and cancer



**Figure 6.** (a-c) Schemes of the various synthesized magnetic nanoparticles including (a) single coated SPIONs, (b) crosslinked coated SPION, and (c) PVA- and crosslinked PEGF- coated CNCs. (d) TEM images of crosslinked PEGF-coated CNCs with various magnifications. (e) MRI image of vials containing different samples with the same iron concentrations (0.02 mg/ml) obtained by Artoscan S.p.A. imager at H=0.2T: (1) Endorem (commercially available dextran coated SPIONs), (2) bare SPIONs, (3) PVA-Coated CNCs, (4) PEGF-coated CNCs, (5) Crosslinked PEGF-Coated CNCs. Release profiles of (f) TMX and (g) DOX from PEGF- and crosslinked PEGF- coated single nanoparticles; and PVA-, PEGF- and crosslinked PEGF- coated CNCs over 300 and 200 hours, respectively. With permission from reference [180].

is employed by FeRx [195]. It is noteworthy to mention that FeRx Inc. was granted fast-track status to proceed with multi-centre Phases I and II clinical trials of their magnetic targeting system for hepatocellular carcinomas. However, in April 2004 FeRx halted its clinical trial, putting into doubt company's ability to continue as a

going concern [2].

### Hyperthermia

One of the crucial capabilities of Magnetic nanoparticles is that they can be made to generate heat, which leads to their use as hyper-

thermia agents, delivering toxic amounts of thermal energy to cancerous tumors where a moderate degree of tissue warming results in more effective cell destruction [202, 203]. The production of local heat in magnetic nanoparticles is due to the fact that the magnetic anisotropy of magnetic nanoparticles can be much greater than those of a bulk materials, while differences in the Curie or Néel temperatures, i.e., the temperatures of spontaneous parallel or antiparallel orientation of spins between magnetic nanoparticles and the corresponding microscopic phases, reach hundreds of degrees [203, 204]. Magnetic nanoparticles were first employed in hyperthermia application in the work of Chan et al. [205] and Jordan et al. [206] in 1993, where they proved that superparamagnetic crystal suspension had great capability to absorb the energy of an alternating magnetic field; this absorbed energy can in turn be converted into heat. Given that tumor cells are more sensitive to a temperature increase than healthy ones [207, 208], this property can be used in vivo to increase the temperature of cancerous tissue and to destroy the pathological cells by hyperthermia.

During the last 2 decades, intense efforts were focused on improvement of hyperthermia techniques for clinical applications. Advances in the area of nanotechnology have contributed to the development of superparamagnetic fluid hyperthermia, which is well recognized as promising method for cancer treatment because of the ease of targeting the cancerous tissue and hence having fewer side effects than chemotherapy and radiotherapy, as proven by the results of current/ongoing clinical trials [209].

Hergt et al. [210] injected 100 mg dextran coated magnetic nanoparticles into the tail vein of Sprague Dawley rats, treated with AC magnetic field (12 min, 450 kHz, unknown field and SAR). According to authors' considerable the tumor shrinkage and tissue necrosis was observed. Jordan and co-workers [211-225] did the first clinical patient trials [226] with magnetic nanoparticles. In this case, a special hyperthermia-generating prototype instrument was developed which is able to generate variable magnetic fields in the range of 0 - 15 kA/m at a frequency of 100 kHz. At the same time, the machine allows for real-time patient temperature measurements to ensure that neither the upper limit of the therapeutic temperature

threshold is exceeded, thus preventing thermal ablation, nor the lower, ineffective limit is crossed. This prototype is capable of treating tumors placed in any region of the body (e.g., prostate cancer, brain tumors).

Maier-Hauff et al. [227] injected neuro-navigationally controlled intra-tumoral instillation of an aqueous dispersion of iron-oxide nanoparticles in 66 patients (59 with recurrent glioblastoma) and subsequently the particles were heated up using an alternating magnetic field. Treatment was combined with fractionated stereotactic radiotherapy. A median dose of 30 grays (Gy) using a fractionation of  $5 \times 2$  Gy/week was applied. The primary study endpoint was overall survival following diagnosis of first tumor recurrence (OS-2), while the secondary endpoint was overall survival after primary tumor diagnosis (OS-1). The median overall survival from diagnosis of the first tumor recurrence among the 59 patients with recurrent glioblastoma was 13.4 months (95% CI: 10.6-16.2 months). Median OS-1 was 23.2 months while the median time interval between primary diagnosis and first tumor recurrence was 8.0 months. The authors claimed that the side effects of their new therapeutic approach were moderate, and no serious complications were observed [227]. According to these results, one can conclude that thermotherapy using magnetic nanoparticles in conjunction with a reduced radiation dose is reasonably safe and effective and leads to longer OS-2 compared to conventional therapies in the treatment of recurrent glioblastoma [227]. Till now, only local hyperthermia is applicable for magnetic fluid hyperthermia; in this case, surface-engineered magnetic nanoparticles, which are dispersed in a carrier fluid, are placed inside the cancerous tumor through direct injection or tumor specific antibody targeting. This specific antibody targeting cause the attachments of nanoparticles to the cancerous cells and the labeled-cells are exposed to an alternating magnetic field. This field makes the magnetic nanoparticles generate heat by magnetic relaxation mechanisms; these local heats can induce apoptosis to the labeled cells.

### *Simultaneous drug delivery and hyperthermia*

One of the crucial problems with magnetic targeted hyperthermia is that a limited dose of nanoparticles reached the tumor tissue. This

resulted in insufficient temperature enhancement in the cancerous sites; thus there is a risk of proliferation of cancer cells that survived during thermotherapy [229]. In order to overcome the problem, several specific tumor receptor targeting moieties together with anticancer drugs can be attached to the surface of particles. These employed targeting moieties together with anti-cancerous drugs may induce cell apoptosis. In this case, the hyperthermia treatment was used as a driving force for simultaneous drug delivery purposes (e.g. using thermosensitive polymers [203], as nanoparticle's coating) [230]. For example,  $\beta$ -cyclodextrin (CD) was used as a drug container for hydrophilic (paclitaxel) or lipophilic (doxorubicin) structures. Drugs incorporated in the CD can thus be released through the use of induction heating, or hyperthermic effects, by applying a high-frequency magnetic field. In this case, folic acid (FA) and CD-functionalized magnetic nanoparticles were synthesized and it was found that by induction of heating, drug release was triggered from the CD cavity on the particle - a behavior that was controlled by switching the high-frequency magnetic field on and off.

Another drug delivery system, based on covalently attaching genistein onto SPIONs coated by cross-linked carboxymethylated chitosan (CMCH), has been developed [231] and the results confirmed that the nanosystem could significantly enhanced cancer cell apoptosis.

### Importance of protein-nanoparticle interactions

It has been long recognized that proteins were associated with nanoparticles, upon entrance of nano-objects into a biological fluid [232-235]. The amount and types of the associated proteins, which is called a protein "corona", on the surface of the nanoparticles leads to an in vivo response [232]. More specifically, composition of the obtained protein corona can be used for prediction of the way cells interact with, recognize and process the nanoparticles [236].

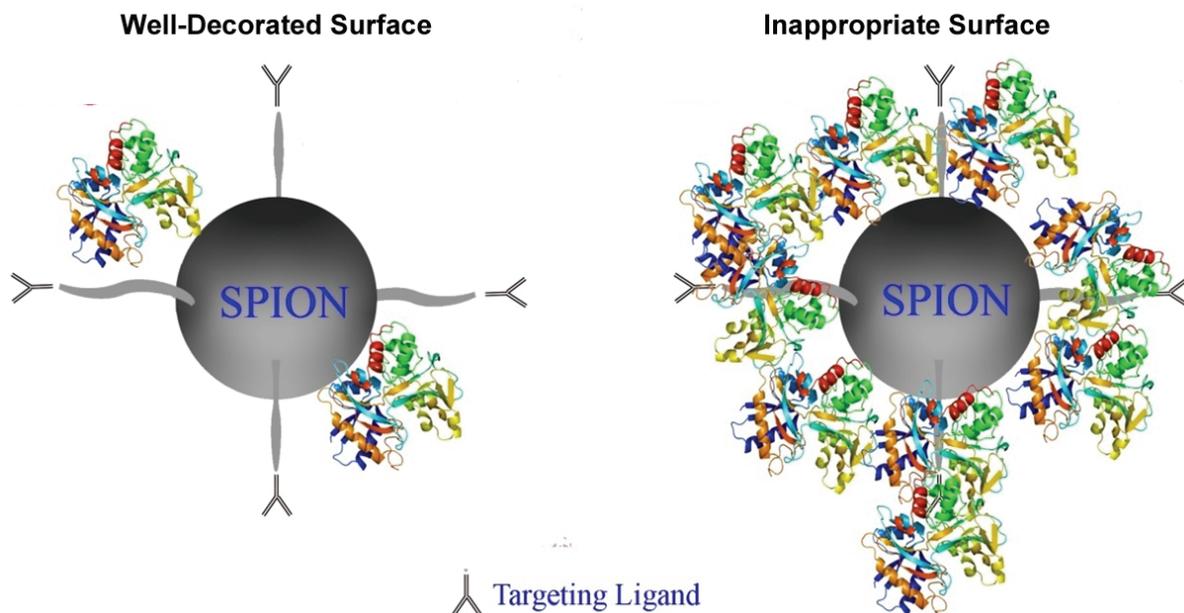
It also has been shown that the composition of the hard corona protein profiles can be significantly different due to the changes in surface chemistries of the same materials [237]. Thus, decoration of the surface of magnetic nanoparticles would be very useful for the efficient design of targeted delivery systems where the targeting moieties are covered on the surface of SPIONs. More specifically, inappropriate surface chemistry of nanoparticles have great capability

to be severely covered by the proteins which may causing the elimination of targeting moieties; in contrast, the targeting species of the well surface decorated particles are active due to their lower protein affinity (lowest thickness of protein corona in comparison with inappropriate surface) to proteins causing the highest targeting yields (see **Figure 7**). Clearly there is a significant amount of work to be done to confirm these issues.

### Conclusions and future perspectives

Multifunctional SPIONs play an important role in the development of simultaneous targeted delivery, imaging, and hyperthermia for diagnosis and treatment of cancerous tumors in vivo. Deeper understanding on the protein corona compositions at the surface of nanoparticles are essential for development of nanoparticle specific uptake by desired cells in vivo. In addition, control of the protein corona which is formed at the surface of magnetic particles, which are decorated by cancer-specific binding agents, would make targeted delivery and magnetic fluid hyperthermia treatment much more selective than traditional chemotherapy and even conventional hyperthermia. Furthermore, multifunctional magnetic particles can be magnetically targeted and concentrated in the target tissue, and drug release together with heating are then only induced to significant burst effects and temperatures where the magnetic nanoparticles have been deposited. In addition, tissue-deposited magnetic particles will generally stay where they were initially deposited, thus allowing for repeated and concentrated drug release and hyperthermia treatments in the same area.

At the moment the amount of particles delivered to the cancerous tissues/cells by means of antibody targeting is too low for a sufficient drug release and temperature increase. Thus, the main challenges in this field will be the design of stealth nanoparticles able to circulate in the blood compartment for a long time and the surface grafting of ligands able to facilitate their specific internalization in cancerous cells. Although the results obtained from the first clinical trials of magnetic nanoparticles are very promising, it would be premature to claim that these molecules contribute therapeutic advantages because survival and disease progression benefits were not defined endpoints of the feasibility studies.



**Figure 7.** Cartoon showing the importance of surface chemistries in targeted delivery/imaging applications.

There are several matters that should be considered to enhance the targeted drug delivery, targeted imaging and hyperthermia yield of SPI-ONs. These include the exact definition of cell membrane composition; in vivo control of drug release; in vivo control of heat distribution; management surface of the nanoparticles' for formation of desired protein corona composition followed by fast internalization by the target cells; and optimization of the biophysicochemical properties of the particles. A non homogeneous particle distribution in the tissue may lead to the occurrence of uncontrollable targeting, drug release, and hot spots where the high drug amounts/temperature could cause non-specific necrosis of the tissue. In contrast, in regions with a low particle concentration the drug and/or temperature would not be sufficiently high enough to trigger the onset of apoptosis and a proliferation of surviving cancerous cells can still occur. Both effects should be prevented and thus more even distribution of particles in the tissue as well as the monitoring of the spatial heat distribution should form the focus of future research.

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